Molecular Simulation of Interactions between Structured Peptides and **Biomaterial Surfaces**

[†]Galen Collier, [‡]Steven J. Stuart, and [†]Robert A. Latour [†]Department of Bioengineering and [‡]Department of Chemistry, Clemson University, Clemson, SC 29634

> AFM image of proteins adsorbed to a surface.

Introduction:

Protein-biomaterial interactions govern the biocompatibility of implanted materials.

Controlling biocompatibility through material design begins with the study of

these interactions at the atomic level, where all-atom molecular dynamics (MD) and replica-exchange (REMD) simulations serve as excellent research tools.

Current MD simulation methods and parameters are not designed to accommodate the unique types of atomic interactions involved in protein-biomaterial interactions, where nonbonded interactions dominate.

The purpose of this work is to evaluate the applicability of existing parameter libraries (CHARMM, AMBER, and OPLS-AA) for representing protein-surface interactions. We are simulating discrete protein structural components in the form of simple model peptide structures (α -helices and β sheets), as they interact with hydrophobic and charged selfassembled monolayer (SAM) surfaces. These simulations are being run as a collaborative effort in parallel with experimental studies (ssNMR, XPS, ToF-SIMS, NEXAFS) of identical peptide-surface systems, which are being conducted by the Castner, Gamble, Stayton, and Drobny groups at the University of Washington.

LKα and LKβ Model Peptides and SAM Surfaces:



Peptides consisting of leucine (L) and lysine (K) serve as excellent peptide models because the periodicity of these amino acids can be arranged to provide distinct regions of hydrophobicity and positive charge, as desired.

LKβ7: (Ac-LKLKLKL-NH₂)



-CH₃ hydrophobic surface



General Computational Approach:

- CHARMM² c34b2 simulation engine.
- CHARMM v27 FF, AMBER94 FF, OPLS-AA FF.
- MMTSB³ toolset for managing REMD simulations.
- 10 ns-scale MD and REMD⁴ simulations.
- Explicit TIP3P water with SHAKE algorithm for fixing X-H's.
- NVT ensemble w/Nosé-Hoover⁵ thermostat.
- New Velocity Verlet⁶ (VV2) integrator.
- Particle-mesh Ewald⁷ (PME) for long-range electrostatics.
- Na⁺ and Cl⁻ ions to achieve ~140 mM saline.
- All REMD runs are prepopulated with a variety of starting conformations (helical, extended, randomized, etc.).

Peptide Stability and Interaction Studies:

MD and REMD simulations of single and multiple peptides with various starting orientations (including random conformations) were used to establish model peptide behavior in solution in the absence of a surface.





Complete Production Systems:

An initial round of MD and REMD simulations of peptides with surfaces has revealed distinct behavioral characteristics of the peptides and of the water structure

over these two different surfaces. (Note low density of water over CH₃-SAM)





Biomolecular Modeling Laboratory





LKB7 pair over hydrophobic -CH₂ SAM

surface (10 ns REMD, CHARMM)

LKa14 over hydrophobic -CH₂ SAM surface (10 ns REMD, CHARMM)





LKa14 over charged -COOH SAM surface (10 ns REMD, CHARMM)





Ramachandran plots characterizing peptide secondary structure (phi & psi dihedral angles of backbone)

Concluding Remarks:

- The LKα14 peptide retains its helical structure in solution and when it is adsorbed to either surface (with adsorption to the CH3 surface predicted to be more stable).
- The pair of LKβ7 peptides maintain an antiparallel configuration in solution and when adsorbed to either surface.
- Adsorption of the charged residues on the oppositely charged surface represents an entropy-driven ion-exchange process in which the multivalent peptide displaces multiple monovalent ions from the surface.

Simulations with the CHARMM force field are providing results that appear to closely represent the realistic behavior of this peptide adsorption system.

Funding by NIH R01 GM074511 and computing resources by TeraGrid grant TG-BCS080002.

- 1) http://nanotech.sc.mahidol.ac.th 2) J. Mol. Graph. Model. 22, 377-395 (2004) 3) Comp. Chem. 4, 187-217 (1983). 4) Chem. Phys. Lett. 314, 141-151 (1999).
- 5) J. Chem. Phys. 119, 3025-3039 (2003). 6) Mol. Phys. 87, 1117 (1996). 7) J. Chem. Phys. 103, 8577 (1995)

-COOH charged surface (50% deprotonated, pKa 7.4)