

## SOLUTION

NAME: \_\_\_\_\_

This is an open book quiz. You may use a four function calculator. An unsigned honors pledge will result in a zero.

1. Consider a Carnot power cycle where the working fluid is 0.5 lb of water. During the isothermal expansion, the water is heated at 600°F from a saturated liquid to a saturated vapor. The water then expands adiabatically until its temperature drops to 90°F and the quality is  $x = 0.643$ . The next process is the isothermal compression. Determine the quality at the end of this isothermal compression.

GIVEN:  $T_H, T_c, m, x_3$ FIND:  $x_4$ ASSUME: No KE or PE effectsANALYSIS:  $\boxed{1 \rightarrow 2}$ 

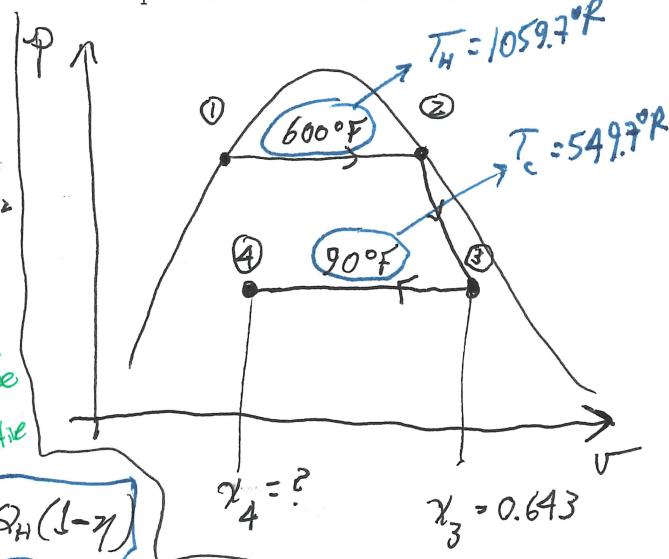
$$\Delta E = Q - W \rightarrow \Delta U = Q - W \rightarrow Q_{12} = \Delta U + W_{12}$$

$$Q_{12} = Q_H = m(u_2 - u_1) + m \int_{p_1}^{p_2} v \, dp = m(u_2 - u_1) + m_p(v_2 - v_1)$$

$$Q_H = m[u_2 + P_2 v_2 - (u_1 + P_1 v_1)] = m(h_2 - h_1) = Q_H$$

$$\eta_{rev} = 1 - \frac{T_c}{T_H} = 1 - \frac{549.7^\circ R}{1059.7^\circ R} = 0.4813 = \eta$$

$$\eta = 1 - \left| \frac{Q_c}{Q_H} \right| \rightarrow \eta - 1 = -\left| \frac{Q_c}{Q_H} \right| \rightarrow 1 - \eta = \left| \frac{Q_c}{Q_H} \right| \rightarrow Q_c = Q_H(1 - \eta)$$



$$\boxed{3 \rightarrow 4} \quad \Delta E = Q - W \quad \Delta U = Q - W \quad Q = \Delta U + W \quad Q_{34} = m(u_4 - u_3) + m \int_{p_3}^{p_4} v \, dp = m(u_4 - u_3) + m_p(v_4 - v_3)$$

$$Q_{34} = m[u_4 - u_3 + P_4 v_4 - P_3 v_3] = m(h_4 - h_3)$$

$$Q_{34} = Q_c = m(h_4 - h_3) \rightarrow Q_c/m = h_4 - h_3$$

$$\text{Combining } h_4 = h_3 + \frac{-Q_H(1-\eta)}{m} = h_3 + \frac{-m(h_2 - h_1)(1-\eta)}{m}$$

$$h_4 = h_3 + (h_2 - h_1)(1-\eta)$$

$$h_2 = h_g(600^\circ F) = 1166.4 \text{ Btu/lb}$$

$$\text{Go to tables } h_1 = h_f(600^\circ F) = 616.7 \text{ Btu/lb} \quad h_3 = (1-x_3)h_f + x_3 h_g$$

$$\text{At } 90^\circ F \quad h_f = 58.07 \text{ Btu/lb} \quad h_g = 1100.7 \text{ Btu/lb} \quad h_3 = 728.48 \text{ Btu/lb}$$

$$\text{So, } h_4 = 728.48 \text{ Btu/lb} + (1166.4 \text{ Btu/lb} - 616.7 \text{ Btu/lb})(1 - 0.4813)$$

$$h_4 = 443.35 \text{ Btu/lb} = (1-x_4)(58.07 \text{ Btu/lb}) + x_4(1100.7 \text{ Btu/lb})$$

$$x_4 = 0.369 \quad \text{ANS.}$$

I HAVE NEITHER PROVIDED OR RECEIVED HELP DURING THIS QUIZ.

SIGNATURE