

SOLUTION

NAME: \_\_\_\_\_

This is an open book quiz. You may use a four function calculator. An unsigned honors pledge will result in a zero.

1. Consider an ideal vapor compression refrigeration cycle. The condenser pressure is 10.0 bar. Saturated vapor enters the compressor, and the compressor exit temperature is 132.865°C. If the working fluid is ammonia, compute the power input to the cycle in units of kJ/kg.

GIVEN: Condenser  $p$ ; "g" state entering compressor; compressor exit  $T$ ;  $\text{NH}_3$ .

FIND:  $\dot{w}/m = ?$  (Note  $\dot{w}_{\text{net}}/m = \dot{w}_c/m$ )

ASSUME: Ideal vapor compression assumptions, namely compressor is isentropic.

ANALYSIS:  $\dot{w}_c/m = h_2 - h_1$

From  $\text{NH}_3$  tables

$$\textcircled{2} \quad p = 10 \text{ bar}, T_2 = 132.865^\circ\text{C}$$

$$\left. \begin{array}{l} h_2 = 1744.54 \text{ kJ/kg} \\ A_2 = 5.8391 \text{ kJ/kg}\cdot\text{K} \end{array} \right\} \begin{array}{l} \text{via} \\ \text{interpolation} \end{array}$$

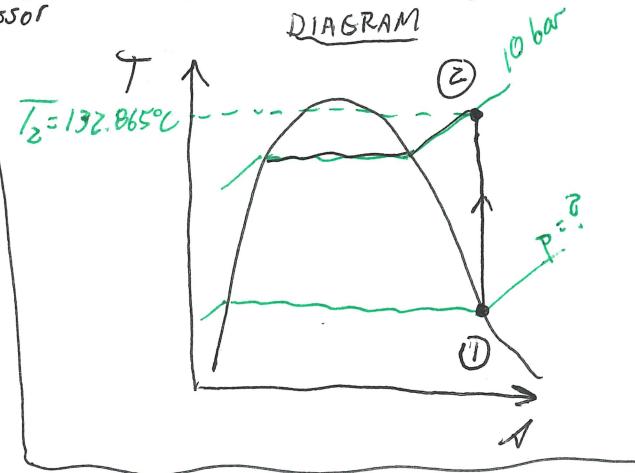
① Go to saturated tables and search

for  $A_g = 5.8391 \frac{\text{kJ}}{\text{kg}\cdot\text{K}}$ . This occurs at  $p = 1 \text{ bar}$

$$\text{where } h_g = 1398.41 \text{ kJ/kg} = h_1$$

$$\text{So, } \dot{w}_c/m = 1744.54 \frac{\text{kJ}}{\text{kg}} - 1398.41 \frac{\text{kJ}}{\text{kg}}$$

$$\boxed{\dot{w}_c/m = 346.13 \frac{\text{kJ}}{\text{kg}}} \quad \leftarrow \text{ANS.}$$



I HAVE NEITHER PROVIDED OR RECEIVED HELP DURING THIS QUIZ.

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SIGNATURE