

SOLUTION

NAME: _____

This is an open book quiz. You may use a 4-function calculator. An unsigned honors pledge will result in a zero.

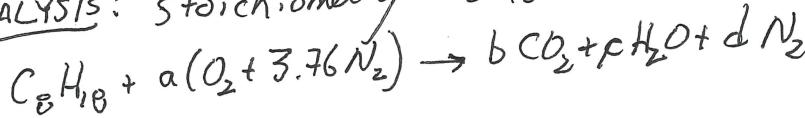
1. Octane vapor and 100% theoretical air enter an uninsulated reactor at SRS. The reactor is sufficiently long, so that the exit is also at SRS. Determine the heat transfer rate in units of kJ per kmole of fuel flowing.

GIVEN: 100% theo air; octane; i, e @ SRS

FIND: $\dot{Q}/n_F = ? \text{ kJ/kmole}$

ASSUME: Usual combustion assumptions. H_2O is vapor, S.S.

ANALYSIS: Stoichiometry C_8H_{18}

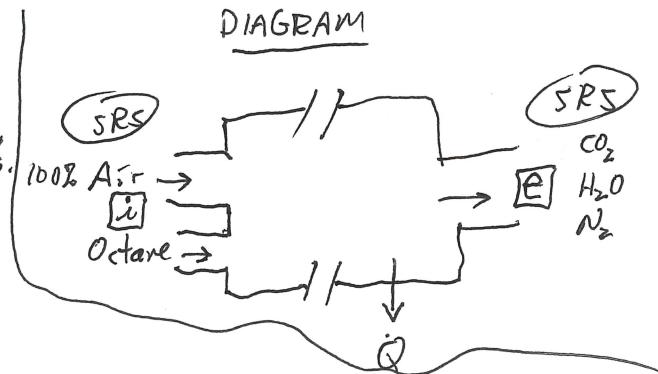
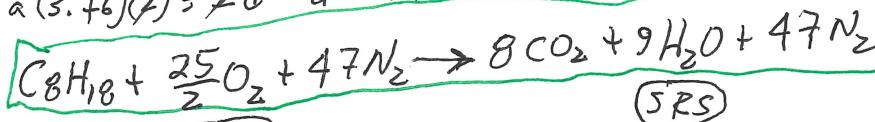


$$C: 8 = b$$

$$H: 18 = 2c \quad c = 9$$

$$O: 2a = 2b + c = 2(8) + 9 = 25 \quad a = 25/2$$

$$N: a(3.76)(2) = 2d \quad d = 47$$



$$\bar{h} = \bar{h}_f + (\bar{h}(T) - \bar{h}(298K))$$

$\Delta h \rightarrow$ since i and e are at SRS ($\Delta \bar{h} = 0$ for everything)

$$\frac{dE}{dt} = \dot{Q} - \dot{n}_F \bar{h}_F + \sum_i n_i \bar{h}_i - \sum_e n_e \bar{h}_e$$

$$\dot{Q} = \sum_e n_e \bar{h}_e - \sum_i n_i \bar{h}_i$$

$$\dot{Q} = [n_{CO_2} \bar{h}_{CO_2} + n_{H_2O} \bar{h}_{H_2O} + n_{N_2} \bar{h}_{N_2}]_{SRS} - [n_F \bar{h}_F + n_{O_2} \bar{h}_{O_2} + n_{N_2} \bar{h}_{N_2}]_{SRS}$$

Divide by n_F

$$\frac{\dot{Q}}{n_F} = 8\bar{h}_{CO_2} + 9\bar{h}_{H_2O} - \bar{h}_F - \frac{25}{2}\bar{h}_{O_2} = 8\bar{h}_{CO_2} + 9\bar{h}_{H_2O} - \bar{h}_{Octane} - \frac{25}{2}\bar{h}_{O_2}$$

$$\frac{\dot{Q}}{n_F} = 8(-393,520 \frac{\text{kJ}}{\text{kmole}}) + 9(-241,820 \frac{\text{kJ}}{\text{kmole}}) - (-208,450 \frac{\text{kJ}}{\text{kmole}})$$

$$\boxed{\frac{\dot{Q}}{n_F} = -5,116,090 \text{ kJ/kmole}} \quad \text{ANS.}$$

I HAVE NEITHER PROVIDED OR RECEIVED HELP DURING THIS QUIZ.

SIGNATURE