

PROBLEM 3.70

KNOWN: Data are provided for water in a piston-cylinder assembly undergoing two processes in series: constant-pressure followed by constant-volume.

FIND: For the overall process find W and Q .

SCHEMATIC & GIVEN DATA:

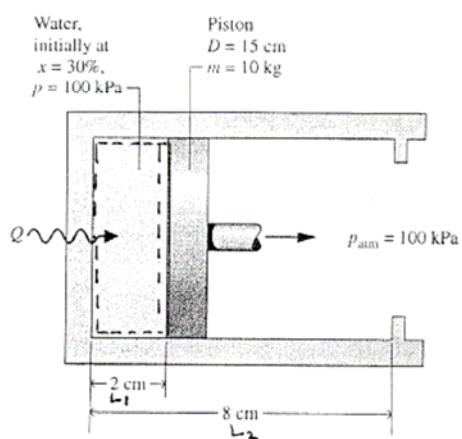


Fig. P3.70

ANALYSIS:

Since volume change is the only work mode, work can be evaluated from Eq. 2.17.

Thus

$$W_{12} = \int p dV = p [V_2 - V_1] \quad (1)$$

Note that $W_{23}=0$ because there is no piston motion (Volume is constant).

An energy balance for the overall process reads.

$$(U_3 - U_1) + \Delta KE + \Delta PE = Q_{13} - W_{12} \Rightarrow Q_{13} = m(u_3 - u_1) + W_{12} \quad (2)$$

Need V_1, V_2, m, u_1, u_3

$$\begin{aligned} \Rightarrow V_1 &= \frac{\pi D^2}{4} L_1 = \frac{\pi (0.15m)^2}{4} (0.02m) \\ &= 3.53 \times 10^{-4} \text{ m}^3 \\ \Rightarrow V_2 &= \frac{\pi D^2}{4} L_2 = \frac{\pi (0.15m)^2}{4} (0.08m) \\ &= 1.414 \times 10^{-3} \text{ m}^3 \end{aligned}$$

$$V_3 = V_2 = \frac{V_2}{m} = \frac{1.414 \times 10^{-3}}{6.94 \times 10^{-4}} = 2.037 \text{ m}^3$$

$$\begin{aligned} \text{With data from Table A-3,} \\ V_f &= V_f + x(V_g - V_f) \\ &= \frac{1.0432}{10^3} + 0.3 \left[\frac{1.694 - 1.0432}{10^3} \right] \\ &= 0.50893 \text{ m}^3/\text{kg} \end{aligned}$$

$$\Rightarrow m = \frac{V_1}{V_f} = 6.94 \times 10^{-4} \text{ kg}$$

$$\begin{aligned} u_1 &= u_f + x(u_g - u_f) \\ &= 417.36 + 0.3(2506.1 - 417.36) \\ &= 1043.98 \text{ kJ/kg} \end{aligned}$$

Interpolating in Table A-4 at 1.5 bars, $u_3 = 2952.1 \frac{\text{kJ}}{\text{kg}}$

Using Eq. (1),

$$W_{12} = p [V_2 - V_1] = (1 \text{ bar}) \left| \frac{10^5 \text{ N/m}^2}{1 \text{ bar}} \right| (14.14 - 3.53) \times 10^{-4} \text{ m}^3 \left| \frac{1 \text{ kJ}}{10^3 \text{ N.m}} \right| \xrightarrow{W} 0.106 \text{ kJ}$$

Using Eq. (2)

$$Q_{13} = m(u_3 - u_1) + W_{12} = (6.94 \times 10^{-4} \text{ kg})(2952.1 - 1043.98) \frac{\text{kJ}}{\text{kg}} + 0.106 \text{ kJ} \xrightarrow{Q} 1.324 + 0.106 = 1.43 \text{ kJ}$$