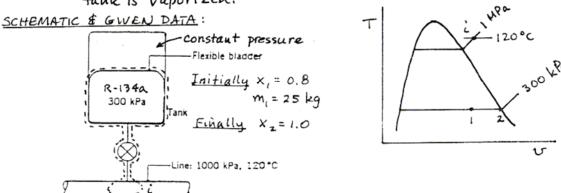
4.118 PROBLEM

A well-insulated tank containing R-134ais connected to a KNOWN: supply line. As refrigerant is allowed to flow into the tank, a flexible bladder in the tank expands to maintain the retrigerant in the tank at constant pressure.

Determine the amount of mass admitted to the tank between FIND: the initial time and the instant when all the liquid in the tank is vaporized.



ENGR. MODEL: (1) The control volume is shown with acr = 0. (2) Conditions in the supply line remain constant. (3) The pressure remains constant in the tank. (4) Kinetic and potential energy effects are negligible.

ANALYSIS: The mass rate balance takes the form; dmcv/dt = mi. With the assumptions listed, the energy rate balance is

The specific enthalpy h; is constant by assumption(z). Thus, combining the mass and energy rate balances and integrating $\Delta U_{cv} = -W_{cv} + \int_{t_1}^{t_2} \dot{m}_i \, h_i \, dt$

$$\Delta U_{cv} = -W_{cv} + \int_{t}^{t} \dot{m}_i h_i dt$$

Since h; is constant
$$\Delta U_{cv} = -W_{cv} + h_i \int_{t_i}^{t_2} \dot{m}_i dt = -W_{cv} + h_i (m_2 - m_1) \qquad (1)$$

To evaluate the work, note that the pressure in the tank is constant. Wcv = SpdV = p(V2-V1) = p(m2 U2-m, U1) (z)

Combining (1) and (2), and noting that AUCr = mzuz-m,u,

$$m_2u_2-m_1u_1 = -p(m_2v_2-m_1v_1) + h_1(m_2-m_1)$$

 $m_2(u_2+pv_1) - h_1 = m_1(u_1+pv_1) - h_1$

 $m_2(u_2+pv_2)-h_1=m_1(u_1+pv_1)-h_1$ 00 m2[h2-hi] = m,[h,-hi]

Continued on next slide

Problem 4-118 continued

Solving for M2

Using data from Table A-11 at 3bar: hg = 50.85, hg = 247.59 10 1/kg

From Table A-12, N; = 356.52 KO/Kg. Thus

$$m_2 = 25 \text{ kg} \left(\frac{208.34 - 356.52}{347.59 - 356.52} \right) = 34.03 \text{ kg}$$

Finally,