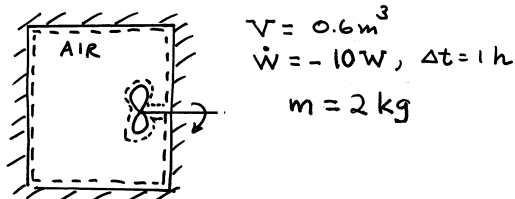


PROBLEM 2.64\*

KNOWN: Air contained in a rigid well-insulated tank receives energy at a specified rate from a paddle wheel.

FIND: Determine the specific volume at the final state, the energy transfer by work, and the change in specific internal energy of the air.

SCHEMATIC & GIVEN DATA:



ASSUMPTIONS: 1. The air is the closed system. 2. For the system,  $Q = 0$  and there are no changes in kinetic energy and potential energy. 3. The initial and final states are equilibrium states.

ANALYSIS: (a) Since the mass and volume are each constant, the specific volume at states 1 and 2 are the same:

$v_2 = v_1$ . Thus

$$v_2 = V/m = (0.6 \text{ m}^3)/(2 \text{ kg}) = 0.3 \frac{\text{m}^3}{\text{kg}}$$

←  $v_2$

(b) To evaluate  $W$ , integrate

$$W = \int_0^{1\text{h}} \dot{W} dt = \int_0^{1\text{h}} (-10 \text{ W}) dt = (-10 \text{ W})(1 \text{ h}) \left| \frac{1 \text{ J/s}}{1 \text{ W}} \right| \left| \frac{3600 \text{ s}}{1 \text{ h}} \right| \left| \frac{\text{kJ}}{10^3 \text{ J}} \right|$$

$$= -36 \text{ kJ}$$

←  $W$

(c) The change in specific internal energy can be found from an energy balance:

$$\cancel{\Delta KE} + \cancel{\Delta PE} + \Delta U = \cancel{Q} - W \Rightarrow m \Delta u = -W$$

or

$$u_2 - u_1 = \frac{(-W)}{m}$$

$$= \frac{-(-36 \text{ kJ})}{2 \text{ kg}} = 18 \frac{\text{kJ}}{\text{kg}}$$

←  $u_2 - u_1$