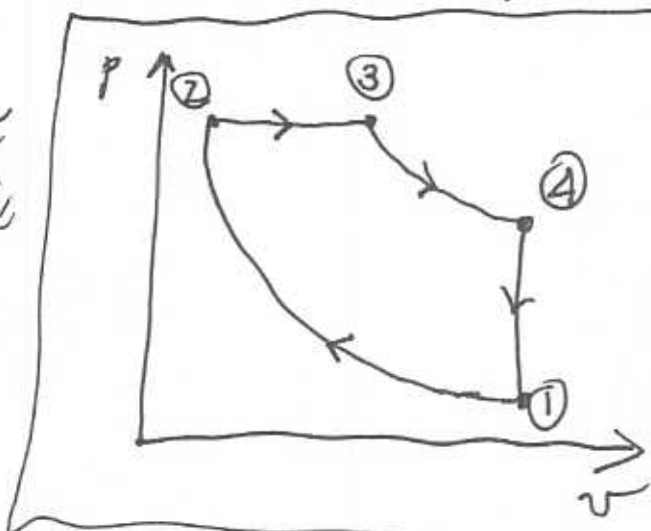


9.20 Air-standard Diesel cycle with constant specific heats (evaluated at 300K)

GIVEN  $P_1 = 95 \text{ kPa}$   $T_1 = 300 \text{ K}$   
 $P_3 = 7.2 \text{ MPa}$   $T_3 = 2150 \text{ K}$



(a)  $r = \frac{v_1}{v_2} = ?$

This is air, an ideal gas, with constant  $(c_p, c_v)$ . Hence we can use Eq. (6.45) - (6.47) for the isentropic processes

$$p v = R T$$

$$v_1 = \frac{R T_1}{P_1} \leftarrow \text{known}$$

Eq. (6.45)  $\frac{T_2}{T_1} = \left( \frac{P_2}{P_1} \right)^{(k-1)/k}$

$$P_2 = P_3 \quad \frac{T_2}{T_1} = \left( \frac{P_3}{P_1} \right)^{\frac{k-1}{k}}$$

Table A-20

$$\left. \begin{array}{l} c_p = 1.005 \text{ kJ/kg}\cdot\text{K} \\ c_v = .718 \text{ kJ/kg}\cdot\text{K} \end{array} \right\} k = 1.400$$

$$\underline{T_2 = 1033 \text{ K}}$$

$$\frac{v_1}{v_2} = \frac{R T_1 / P_1}{R T_2 / P_2} = \frac{T_1 / P_1}{T_2 / P_3}$$

$$\boxed{r = 22.01}$$

9.20 (b) The cutoff ratio  $r_c = \frac{v_3}{v_2}$

$$v_3 = \frac{RT_3}{P_3} \quad v_2 = \frac{RT_2}{P_2}$$

$$r_c = \frac{T_3/P_3}{T_2/P_2} = \frac{T_3}{T_2} = \underline{\underline{2.08}}$$

(c)  $\eta = ?$   $\eta = \frac{W_{cyc}/m}{Q_{23}/m} = 1 - \left( \frac{u_4 - u_1}{h_3 - h_2} \right)$

$$\eta = 1 - \frac{c_v(T_4 - T_1)}{c_p(T_3 - T_2)} = 1 - \frac{(T_4 - T_1)}{k(T_3 - T_2)}$$

known  $\swarrow$   $\nwarrow$  known

$$\left[ \frac{T_4}{T_3} \right] = \left( \frac{v_3}{v_4} \right)^{k-1}$$

$$\frac{v_3}{v_4} = \frac{v_3}{v_1}$$

$$\left. \begin{aligned} r &= \frac{v_1}{v_2} \\ r_c &= \frac{v_3}{v_2} \end{aligned} \right\}$$

$$\frac{v_3}{v_1} = \frac{r_c}{r} = \frac{2.08}{22.01} = 0.0946$$

$$\frac{T_4}{T_3} = (0.0946)^{1.40-1} = .389$$

$$T_4 = 837K$$

$$\eta = 1 - \frac{1}{1.40} \left[ \frac{837K - 300K}{2150K - 1033K} \right] = \underline{\underline{0.657}}$$



9.20(d) continued

$$v_1 = \frac{RT_1}{P_1} = \frac{(0.287 \frac{\text{kJ}}{\text{kg} \cdot \text{K}})(300\text{K})}{95 \text{ kPa}} = 0.906 \text{ m}^3/\text{kg}$$

$$mep = \frac{(1.005 \frac{\text{kJ}}{\text{kg} \cdot \text{K}})(2150\text{K} - 1033\text{K}) - (0.718 \frac{\text{kJ}}{\text{kg} \cdot \text{K}})(837\text{K} - 300\text{K})}{(0.906 \text{ m}^3/\text{kg})(1 - \frac{1}{22.01})}$$

$$mep = \frac{737 \text{ kJ/kg}}{0.865 \text{ m}^3/\text{kg}}$$

$$mep = 852 \text{ kPa}$$